

TRITERPENES FROM *GANODERMA LUCIDUM*

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(Revised received 27 November 1987)

Key Word Index—*Ganoderma lucidum*; Polyporaceae; lanostane triterpenes.

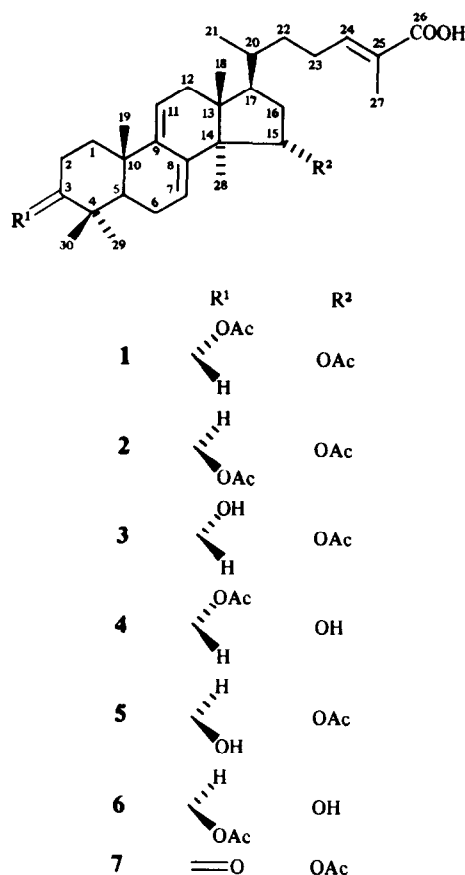
Abstract—Three new lanostanoid triterpenes, isolated from the mycelia of the fungus *Ganoderma lucidum*, were determined to be 3 β -hydroxy-15 α -acetoxy-lanosta-7,9(11),24-trien-26-oic acid, 3 β -acetoxy-15 α -hydroxy-lanosta-7,9(11),24-trien-26-oic acid and 3-oxo-15 α -acetoxy-lanosta-7,9(11),24-trien-26-oic acid by spectroscopic and chemical methods.

INTRODUCTION

Ganoderma lucidum (Fr.) Karst, a Polyporaceae species used in traditional Chinese medicine, has attracted great attention recently because of its production of many biologically active oxygenated triterpenes [1–4]. As a continuation of our study on their hypocholesterolemic activities and the previous identification of 3 α ,15 α -diacetoxy-lanosta-7,9(11),24-trien-26-oic acid (ganodermic acid R) (1) and 3 β ,15 α -diacetoxy-lanosta-7,9(11),24-trien-26-oic acid (ganodermic acid S) (2) [5], five more lanostanoid acids 3–7 were isolated. Among them, only 3 (ganoderic acid X) and 4 (ganoderic acid Mf) were previously reported [1, 6]. New compounds 5, 6, and 7 were tentatively named as ganodermic acids T-N, T-O and T-Q, respectively.

RESULTS AND DISCUSSION

The UV spectra (in MeOH) of compounds 3–7 showed almost identical absorption bands at 235, 243, and 252 nm, indicating the presence of a common heteronuclear conjugated diene skeleton. Identical molecular ion peaks at m/z 512 ($C_{32}H_{48}O_5$) (EIMS, 12 eV) and two common fragment ions at 494 [$M - H_2O$] $^+$ and 452 [$M - MeCO_2H$] $^+$ observed in compounds 3–6 suggested that they were stereo- or positional isomers and each compound possessed one hydroxy and one acetoxy substituents in the molecule. Furthermore, two fragment ions at m/z 311 [$M - MeCO_2H - C_8H_{13}O_2$ side chain] $^+$ and 257 (D-ring cleavage—Me) in 3 and 5, which were not observed in 4 and 6, indicated that 3 and 5 bear the acetoxy group at D-ring. The 1H NMR spectra of 3–6 each showed a chemical shift at δ 2.02–2.06 (s, 3H) (Table 1) confirming the presence of an acetoxy group. Comparison of the identical proton signals at δ 5.04 (*dd*) in 3 and 5 with those of ganodermic acids R (1), S (2) and related triterpenoid metabolites [3, 7, 8] indicated that the acetoxy moieties were most likely attached to C-15 with the α -configuration. Compound 3 has a proton signal at δ 3.43 (*br s*) and compound 5 a signal at 3.23 (*dd*) which supported the location of their hydroxy groups at C-3, with the α -configuration in 3 and the β -configuration in



5. Acetylation of 3 and 4 afforded 1 finally confirming 3 as the reported 3 α -hydroxy-15 α -acetoxy-lanosta-7,9(11),24-trien-26-oic acid (ganoderic acid X) (mp 104–105°) [1] and 4 as its positional isomer. Compound 4 is therefore assigned the structure 3 α -acetoxy-15 α -hydroxy-lanosta-7,9(11),24-trien-26-oic acid (ganoderic acid Mf) [6].

Table 1. Partial ^1H NMR spectral data of compounds 3–7 (400 MHz, CDCl_3)*

C	3	4	5	6	7
H-1	†	†	†	†	2.75 <i>ddd</i> (5.7, 14.6, 14.6)
H-3	3.43 <i>s</i>	4.65 <i>s</i>	3.23 <i>dd</i> (5.1, 10.2)	4.48 <i>dd</i> (5.2, 10.0)	—
H-7	5.46 <i>m</i>	5.83 <i>m</i>	5.47 <i>m</i>	5.83 <i>m</i>	5.51 <i>m</i>
H-11	5.32 <i>d</i> (5.9)	5.30 <i>d</i> (5.3)	5.29 <i>d</i> (4.5)	5.28 <i>d</i> (4.7)	5.37 <i>d</i> (5.6)
H-15	5.04 <i>dd</i> (4.9, 9.2)	4.27 <i>dd</i> (5.2, 9.4)	5.04 <i>dd</i> (4.7, 9.4)	4.25 <i>dd</i> (5.1, 9.0)	5.06 <i>dd</i> (4.6, 9.4)
3H-18	0.64 <i>s</i>	0.60 <i>s</i>	0.63 <i>s</i>	0.58 <i>s</i>	0.66 <i>s</i>
3H-19	0.96 <i>s</i> †	0.96 <i>s</i> †	0.98 <i>s</i> †	0.97 <i>s</i> †	1.09 <i>s</i> †
3H-21	0.90 <i>d</i> (6.2)	0.89 <i>d</i> (6.2)	0.89 <i>d</i> (6.2)	†	0.90 <i>d</i> (6.3)
H-24	6.84 <i>t</i> (7.0)	6.86 <i>t</i> (6.8)	6.83 <i>t</i> (7.1)	6.85 <i>t</i> (7.0)	6.82 <i>t</i> (6.9)
3H-27	1.81 <i>s</i>	1.82 <i>s</i>	1.80 <i>s</i>	1.80 <i>s</i>	1.81 <i>s</i>
3H-28	0.99 <i>s</i> †	0.98 <i>s</i> †	0.98 <i>s</i> †	0.90 <i>s</i> †	0.98 <i>s</i> †
3H-29	0.91 <i>s</i> †	0.85 <i>s</i> †	0.85 <i>s</i> †	0.86 <i>s</i> †	1.07 <i>s</i> †
3H-30	0.96 <i>s</i> †	0.96 <i>s</i> †	0.94 <i>s</i> †	0.92 <i>s</i> †	1.16 <i>s</i> †
OAc	2.06 <i>s</i>	2.03 <i>s</i>	2.06 <i>s</i>	2.02 <i>s</i>	2.07 <i>s</i>

*Values in parentheses are coupling constants in Hz.

†Overlapped with other signals.

‡Signals were tentatively assigned.

Ganodermic acids T-N (**5**) and T-O (**6**), which upon acetylation both gave **2**, were also positional isomers. The attachment of an acetoxy substituent in **5** was deduced from the mass spectral data. Compound **5** gave a proton signal at δ 3.23 (*dd*) which changed to 4.48 (*dd*) in **6** and both showed strong coupling with their C-2 protons. This suggested that their C-3 substituents were both in the β -configuration, with a hydroxy group in **5** and an acetoxy group in **6**. The close similarity of the ^{13}C NMR spectra between **5** and **3**, except for a characteristic downfield chemical shift pattern of **5** in the A-ring and particularly at C-3, also revealed that the hydroxy group of **5** was at the C-3 β position (Table 2). Compound **5** was therefore delineated as 3 β -hydroxy-15 α -acetoxy-lanosta-7,9(11),24-trien-26-oic acid (ganodermic acid T-N) (prefixed with a T for Taipei) (mp 145–146 $^\circ$). Comparison of spectral data of **6** with those of **3**, **4** and **5** makes its structure become obvious. Compound **6** is concluded to be the epimer of **4** at C-3 and positional isomer of **5** at C-3/C-15. Chemical transformation and spectral analysis lead us to the structure of **6** as 3 β -acetoxy-15 α -hydroxy-lanosta-7,9(11),24-trien-26-oic acid (ganodermic acid T-O) (mp 160–162 $^\circ$).

Compound **7** is a minor component. It showed a prominent molecular ion peak at m/z 510 (100, $\text{C}_{32}\text{H}_{46}\text{O}_5$) and two fragment ions at m/z 450 [$\text{M} - \text{MeCO}_2\text{H}$] $^+$ and 255 (D-ring cleavage-Me). ^1H NMR spectral data showed a proton signal at δ 2.75 (*ddd*) for H-1 β and a distinct ^{13}C signal at δ 216.59 for a carbonyl, which are known to be characteristic of the 3-oxo-lanostanoid triterpenes [7, 8]. Oxidation of **5** with Jones reagent to yield **7** confirmed that **7** was 3-oxo-15 α -acetoxy-lanosta-7,9(11),24-trien-26-oic acid (ganodermic acid T-

Q). The natural occurrence of oxygenated triterpenoid metabolites in *G. lucidum* as stereo- and positional isomers appears very interesting from the biosynthetic point of view.

In the course of our research on hypocholesterolemic constituents of *G. lucidum* we found that several triterpene isolates including compounds **4** and **6** exhibited an inhibitory activity on cholesterol synthesis.

EXPERIMENTAL

The extraction and fractionation of the mycelia of *G. lucidum* have been described [5]. Pooled EtOAc fractions were chromatographed on a silica gel column (45 \times 2.5 cm) by stepwise elution with increasing percentage of MeOH in CHCl_3 . Fractions containing compounds **3**–**7** were combined and chromatographed by reversed phase high performance TLC (E. Merck HPTLC RP-18, F₂₅₄; 0.25 mm thickness; 10 \times 10 cm; MeCN–HOAc, 100:0.1). Eluting the band at R_f 0.28 with MeOH yielded resinous compound **7** (2.0 mg). Eluting the band at R_f 0.23 with the same solvent gave a mixture of **3** and **4** and eluting the band at R_f 0.20 afforded a mixture of **5** and **6**. Complete separation of **3** and **4** was achieved by normal phase TLC (Merck Kieselgel 60 F₂₅₄; 0.25 mm thickness) using *n*-hexane–Et₂O–EtOAc–HOAc (400:200:200:1, developed \times 3). After elution the bands at R_f 0.24 and 0.19 separately with 5% MeOH in CHCl_3 compounds **3** (4.7 mg) and **4** (10.3 mg) were obtained, respectively. Resolution of **5** and **6** was achieved by normal phase TLC using CHCl_3 –Et₂O–HOAc (380:20:1, developed three times). The bands at R_f 0.23 and 0.17 were eluted separately with 5% MeOH in CHCl_3 to afford compounds **5** (14.4 mg) and **6** (24.7 mg), respectively.

Table 2. ^{13}C NMR spectral data of compounds 3–7 (50.3 or 100.6 MHz, CDCl_3)

C	3	4*	5*	6*	7
1	29.78 <i>t</i>	30.53 <i>t</i>	35.61 <i>t</i>	35.28 <i>t</i>	36.60 <i>t</i>
2	25.44 <i>t</i>	23.04 <i>t</i>	27.62 <i>t</i>	24.09 <i>t</i>	37.48 <i>t</i>
3	75.97 <i>d</i>	78.00 <i>d</i>	78.80 <i>d</i>	80.66 <i>d</i>	216.59 <i>s</i>
4	37.20 <i>s</i>	36.41 <i>s</i>	38.54 <i>s</i>	37.43 <i>s</i>	47.42 <i>s</i>
5	42.81 <i>d</i>	43.93 <i>d</i>	48.74 <i>d</i>	48.94 <i>d</i>	50.39 <i>d</i>
6	22.85 <i>t</i>	22.69 <i>t</i>	22.88 <i>t</i>	22.63 <i>t</i>	23.63 <i>t</i>
7	121.17 <i>d</i>	121.10 <i>d</i>	121.28 <i>d</i>	121.02 <i>d</i>	121.04 <i>d</i>
8	140.08 <i>s</i>	140.66 <i>s</i>	140.04 <i>s</i>	140.64 <i>s</i>	140.37 <i>s</i>
9	145.86 <i>s</i>	145.98 <i>s</i>	145.83 <i>s</i>	145.73 <i>s</i>	145.00 <i>s</i>
10	37.23 <i>s</i>	37.21 <i>s</i>	37.33 <i>s</i>	37.15 <i>s</i>	37.25 <i>s</i>
11	115.45 <i>d</i>	115.52 <i>d</i>	115.76 <i>d</i>	116.01 <i>d</i>	116.92 <i>d</i>
12	37.85 <i>t</i>	38.34 <i>t</i>	37.89 <i>t</i>	38.34 <i>t</i>	37.99 <i>t</i>
13	43.99 <i>s</i>	44.33 <i>s</i>	43.99 <i>s</i>	44.21 <i>s</i>	44.06 <i>s</i>
14	51.29 <i>s</i>	51.95 <i>s</i>	51.19 <i>s</i>	51.83 <i>s</i>	51.31 <i>s</i>
15	77.29 <i>d</i>	74.57 <i>d</i>	77.25 <i>d</i>	74.48 <i>d</i>	77.23 <i>d</i>
16	36.91 <i>t</i>	39.89 <i>t</i>	36.89 <i>t</i>	39.77 <i>t</i>	36.97 <i>t</i>
17	48.74 <i>d</i>	48.71 <i>d</i>	48.74 <i>d</i>	48.67 <i>d</i>	48.85 <i>d</i>
18	15.85 <i>q</i>	15.84 <i>q</i>	15.85 <i>q</i>	15.82 <i>q</i>	16.00 <i>q</i>
19	22.66 <i>q</i>	22.56 <i>q</i>	22.71 <i>q</i>	22.74 <i>q</i>	22.44 <i>q</i>
20	35.85 <i>d</i>	35.81 <i>d</i>	35.84 <i>d</i>	35.76 <i>d</i>	35.92 <i>d</i>
21	18.06 <i>q</i>	18.17 <i>q</i>	18.09 <i>q</i>	18.12 <i>q</i>	18.16 <i>q</i>
22	34.55 <i>t</i>	34.62 <i>t</i>	34.54 <i>t</i>	34.67 <i>t</i>	34.62 <i>t</i>
23	25.83 <i>t</i>	25.75 <i>t</i>	25.83 <i>t</i>	25.70 <i>t</i>	25.92 <i>t</i>
24	145.00 <i>d</i>	145.13 <i>d</i>	144.94 <i>d</i>	145.01 <i>d</i>	144.53 <i>d</i>
25	126.64 <i>s</i>	126.77 <i>s</i>	126.74 <i>s</i>	126.81 <i>s</i>	126.76 <i>s</i>
26	172.52 <i>s</i>	172.86 <i>s</i>	172.86 <i>s</i>	172.93 <i>s</i>	172.10 <i>s</i>
27	11.88 <i>q</i>	11.89 <i>q</i>	11.89 <i>q</i>	11.86 <i>q</i>	12.04 <i>q</i>
28	18.39 <i>q</i>	17.13 <i>q</i>	18.27 <i>q</i>	16.99 <i>q</i>	18.24 <i>q</i>
29	28.07 <i>q</i>	27.66 <i>q</i>	28.06 <i>q</i>	27.95 <i>q</i>	25.40 <i>q</i>
30	22.55 <i>q</i>	22.35 <i>q</i>	15.70 <i>q</i>	16.81 <i>q</i>	22.14 <i>q</i>
AcCO	171.12 <i>s</i>	170.75 <i>s</i>	171.07 <i>s</i>	170.90 <i>s</i>	171.21 <i>s</i>
AcMe	21.31 <i>q</i>	21.19 <i>q</i>	21.30 <i>q</i>	21.16 <i>q</i>	21.40 <i>q</i>

*Spectra were obtained at 50.3 MHz.

Acknowledgements—Support of this work by the National Science Council and Veterans General Hospital, ROC to M.-S. Shiao is gratefully acknowledged.

REFERENCES

- Toth, J. O., Luu, B. and Ourisson G. (1983) *J. Chem. Res. (M)*, 2722.
- Kohda, H., Tokumoto, W., Sakamoto, K., Fujii, M., Hirai, Y., Yamasaki, K., Komoda, Y., Nakamura, H., Ishihara, S. and Uchida, M. (1985) *Chem. Pharm. Bull.* **33**, 1367.
- Morigiwa, A., Kitabatake, K., Fujimoto, Y. and Ikegawa, N. (1986) *Chem. Pharm. Bull.* **34**, 3025.
- Hirofani, M., Ino, C., Furuya, T. and Shiro, M. (1986) *Chem. Pharm. Bull.* **34**, 2282.
- Shiao, M.-S., Lin, L.-J., Yeh, S.-F. and Chou, C.-S. (1987) *J. Nat. Prod.* **50**, 886.
- Nishitoba, T., Sato, H., Shirasu, S. and Sakamura, S. (1987) *Agric. Biol. Chem.* **51**, 619.
- Kubota, T., Asaka, Y., Miura, I. and Mori, H. (1982) *Helv. Chim. Acta.* **65**, 611.
- Hirofani, M., Furuya, T. and Shiro, M. (1985) *Phytochemistry* **24**, 2055.